The Location of the Mercury Atoms in Crystals of Ribonuclease-Parachloromercuribenzoate

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The positions of the mercury atoms in ribonuclease-parachloromercuribenzoate (Ri-pcmb) have been determined from a study of Patterson difference projections on (010) and sharpened Patterson sections. This has permitted the calculation of the structure factors of the mercury atoms, leading to the synthesis of Fourier series for the metal-free crystal.

1. Introduction

Bernal, Carlisle and Rosemever (1959) showed that when crystals of ribonuclease-parachloromercuribenzoate containing chemically 1 atom of Hg per protein molecule (Dix's crystals) were soaked in 60% aqueous alcohol solutions containing 0.5% Na-pcmb, the symmetry of the crystals was unaffected. This diffusion process gave approximately 2 Hg atoms per protein molecule, as chemically determined by Rosemeyer. There was some evidence, from a comparison of the Patterson sections at $v = \frac{1}{2}$ for the soaked Dix's crystal and the metal-free one, that the molecules of one had moved slightly with respect to the other, showing that the crystals were not strictly isomorphous. We therefore decided to work on the soaked Dix's crystals because of the increased intensity changes in the X-ray reflections with respect to those from the metal-free one. In what follows we shall show that in the soaked Dix crystals the Hg atoms occupy two sites per protein molecule and that in the Dix crystal itself these sites are statistically filled. Since the work on the former was carried out before that on the latter, it is convenient to retain this order in the present account.

2. Location of the Hg atoms in the soaked Dix crystal

Fig. 1(a) shows a Patterson difference map on (010) (Bernal, Carlisle and Rosemeyer, 1959) using $(\Delta |F(h0l)|)^2$ between the soaked-Dix and normal crystal, incorporating reflections to spacings of 2Å. This map was not interpretable. We next followed the procedure of Green, Ingram, and Perutz (1954) using a smoothing function of the type exp $[-B\sin^3\theta]$, to limit the reflections to spacings of 6Å; the resulting map is shown in Fig. 1(b). This map can be interpreted in terms of the vector relation $(A+B)^2 = A^2 + 2AB + B^2$ relating 2 pairs of atoms through a centre, which is shown graphically by the parallelogram inscribed on it. The peaks 2AB, showing vectors between atoms not related by the centre, must fall half way between the peaks A^2 and B^2 . The additional peak seen on this map cannot be included in this scheme and is not therefore associated with a Hg-Hg vector but is more likely to be due to a relative movement of the molecules between the two crystals.

Fig. 1(b) can only give very approximate x and z coordinates for the Hg atoms occupying sites A and B. Furthermore it is necessary to know the relative distance Δy of site A with respect to B along the b axis of the crystal (space group $P2_1$) and for this purpose we were strongly advised by Professor D. M. Hodgkin, F.R.S., to look at the individual fully-sharpened three-dimensional Pattersons of the two crystals. These calculations were very kindly carried out for us by Professor K. N. Trueblood on the Electronic Computer S.W.A.C. of the University of California.

The coefficients of the two Patterson series, each containing about 7,000 terms and scaled approximately with respect to each other, were sharpened by dividing each term by an average f^2 (averaged according to atomic percentages), multiplying by exp. $[3B \sin^2\theta/\lambda^2]$, followed by modification with a function

$$\frac{1}{2}\sin^4\theta \cdot \exp\left[-50\sin^2\theta/\lambda^2\right]$$

which reduces the effects of the low order terms and enhances those in the region of 3A. The functions were calculated at 1/60 intervals of the cell edges, for the $\frac{1}{4}$ unit cell, on sections perpendicular to b, which correspond to intervals of approximately 0.5, 0.65, and 0.9 Å along a, b and c respectively. The origin peak was also removed. For economy of space and for purposes of the argument only sections v=6/60, 24/60, and 30/60 for the two crystals have been reproduced (Figs. 2 and 3).

In the space groups $P2_1$, the equivalent points are (x, y, z) and $(\bar{x}, y + \frac{1}{2}, \bar{z})$. For 2 Hg atoms per asymmetric unit, the equivalent points are:

Sites A,
$$x_1, y_1, z_1$$
 and $\bar{x}_1, y_1 + \frac{1}{2}, \bar{z}_1$
Sites B, x_2, y_2, z_2 and $\bar{x}_2, y_2 + \frac{1}{2}, \bar{z}_2$

Excluding the vectors at the origin, the 12 remaining vectors between these sites are of two types:



Fig. 1.

(1) Those lying on $v = \frac{1}{2}$ which may be called the A^2 and B^2 type, have components

 (2) Those of the AB type situated on the planes $v = \pm (y_1 - y_2)$ and $\pm (y_1 - y_2) - \frac{1}{2}$ having components:

$$\begin{array}{l} (x_1 - x_2), \ \pm (y_1 - y_2), \ (z_1 - z_2); \ (\overline{x_1 - x_2}), \ \pm (y_1 - y_2), \\ (\overline{z_1 - z_2}) \\ \text{and} \ (x_1 + x_2), \ \pm \{(y_1 - y_2) - \frac{1}{2}\}, \ (z_1 + z_2) \\ \text{and} \ (\overline{x_1 + x_2}), \ \pm \{(y_1 - y_2) - \frac{1}{2}\}, \ (\overline{z_1 + z_2}) \ . \end{array}$$

The AB type of vectors, owing to Patterson symmetry, have mirror image relationships to each other about v=0 and $v=\frac{1}{2}$; hence the appearance of vectors of the type 2AB when seen in projection on (010), see Fig. 1(b).

This means that the comparisons of the Patterson sections of the crystals, with and without the metal atoms, must be done by examining pairs of planes, e.g. v=1/60 and 29/60, 2/60 and 28/60, and so on to find sufficiently heavy peaks of the AB type, of the same height, whose vector relationships are consistent with the appearence of the A^2 and B^2 peaks on $v=\frac{1}{2}$. It is advisable to begin the interpretation by looking for the former vectors rather than the latter because of the large number of peaks between related atoms





Fig. 1. Patterson difference maps projected on (010) for: (a) Soaked Dix crystal-normal crystal, using reflections with d>2 Å. (b) Soaked Dix crystal-normal crystal, using smoothing function to terminate series at d=6 Å. (c) Dix crystal-normal crystal, using smoothing function to terminate series at d=6 Å.

(there being about 1000 in each molecule) which would fall on $v = \frac{1}{2}$ and so make interpretation difficult. Added to this there must also be a large number of non-Harker peaks falling on this plane, which would tend to reduce the effects of the A^2 , B^2 peaks of the Hg-Hg vectors.

If one accepts the interpretation of the Patterson difference map, Fig. 1(b), as reasonably correct, then it is only necessary to look in the 2AB regions of the two 3-dimensional Pattersons along b on pairs of planes equidistant from v=0 and $\frac{1}{2}$ to find the AB vectors. This was not done. Instead, we took about half the magnitude of the average peak height for the Ri-pcmb Patterson function as a lower limit and any peak below this was rejected. Then we drew up a list of all peaks above this height which appeared on the Ri-pcmb Patterson sections but not on the corresponding Ri-sections. Fig. 4 shows a plot of these peak positions on the Patterson difference map; the number beside each position corresponds to the section v on which it was located. For any given pair of planes, e.g., v=1/60, 29/60, we took all combinations of these ABvectors, determined the x and z co-ordinates of each A and B position and attempted to locate the A^2 , B^2 vectors on $v=\frac{1}{2}$, by comparison of the corresponding Patterson maps. (Figs. 3 (a) and (b)). Out of the many pairs of AB vectors that arose, there was only one that yielded vectors A^2 and B^2 that fell on positive areas on $v=\frac{1}{2}$ for the Ri-*pcmb* Patterson. In every other case either A^2 or B^2 (and sometimes both vectors) fell in negative areas on this section, thus providing sufficiently stringent conditions for their rejection.

The arrows on Figs. 2(a) and (b) indicate the AB peaks on the Ri-*pcmb* Patterson sections v=6/60 and 24/60 that were finally selected, and Figs. 2(c) and (d) show the corresponding sections for the ribonuclease crystal. The two indicated peaks on Figs. 2(a) and (b)



Fig. 2. Fully-sharpened Patterson sections: (a) Soaked Dix crystal, v=6/60. (b) Normal crystal, v=6/60. (c) Soaked Dix crystal, v=24/60. (d) Normal crystal, v=24/60. The arrows indicated the positions of the AB peaks of the mercury atoms in the soaked-Dix crystal.

are of the same height (and their positions have also been indicated by crosses on Fig. 4). Though they are by no means the highest in the functions, they lie well above the lower limit set in our investigation.

Figs. 3(a) and (b) show the section $v = \frac{1}{2}$ for the soaked Ri-*pcmb* and ribonuclease crystals respectively. The comparison of these two sections by themselves shows how difficult it would have been to identify the A^2 , B^2 vectors which are shown by the arrows on Fig. 3(a). The interesting and important finding is that the sharpening of the two Patterson functions has revealed the effect of the Hg-Hg vectors on this section, and for comparison we reproduce the corresponding unsharpened sections of the two crystals, Figs. 3(c) and (d). It is easily seen what sharpening has accomplished.

The AB vector on v=6/30 and the other on 24/30 for the Ri-*pcmb* crystal give more reliable x and z co-ordinates for the Hg atoms than those that would have been obtained from Fig. 1(b). Using these vector relationships, the two Hg atoms in the asymmetric unit have the following co-ordinates:

	\boldsymbol{x}	y	z
\boldsymbol{A}	0.123	0.049	0.075
B	0.806	0.951	0.175

The y coordinates of the Hg atoms are only relative to one another and $\Delta y = 0.098 \simeq 6/60 = 1/10$. It is of interest in this connection to compare the 0k0 reflections of the Ri- and Ri-*pcmb* crystals. This is shown in Fig. 5. The biggest changes are at 040, 0,10,0, and 0,12,0. The reversal of the distribution of intensity in



Fig. 3. Fully-sharpened Patterson sections: (a) Soaked Dix crystal, v=30/60. (b) Normal crystal, v=30/60 compared with the corresponding unsharpened Patterson sections: (c) Soaked Dix crystal, v=30/60. (d) Normal crystal, v=30/60. The arrows indicate the positions of the A^2 and B^2 peaks of the mercury atoms in the soaked Dix crystal.

going from ribonuclease to the soaked Ri-*pcmb* crystal for 0,10,0 and 0,12,0 is consistent with the Δy value found from the 3-dimensional Pattersons, and the change in 040 is due to the fact that the two Hg atoms have such a small Δy separation, and must be lying on planes approximately between the two molecules.

Table 1, column 5, shows the calculated F(h0l)structure factors for the two Hg atoms to spacings of 2.5 Å. Column 4 shows the $|\Delta F(h0l)|$ of the crystals with and without the heavy atoms. As expected, the overall agreement is not very good, indicating that the relative movement of the molecules in the two crystals is possibly responsible for the discrepancies between the amplitudes of the corresponding reflections in these two columns. Some indication that this is likely to be so is shown by Fig. 6.

3. The location of the Hg atoms in the unsoaked Dix's crystal from Patterson-difference maps on (010)

Dix crystals, as stated earlier, contain 1 Hg atom per protein molecule. Since the soaked Dix crystals contain two heavy-atom sites per molecule, it is not unreasonable to suggest that the Hg atom in Dix crystal is statistically divided between the two positions. The intensities of the X-ray reflections from ribonuclease and soaked Ri-*pcmb* crystals were visually estimated. As the intensity differences of the corresponding reflections between the unsoaked Dix crystal and ribonuclease itself are smaller than the intensity differences between the soaked Dix crystal and ribonuclease for the same reflections, it is necessary

 Table 1. The calculated structure factors for the Hg atoms in the soaked Ri-pcmb crystals compared with the observed differences in structure amplitudes between the crystals with and without the metal atoms.

l	$ F_{\mathbf{Ri}} $	$ F_{\rm RiHg} $	$\varDelta F$	$4F_{\mathrm{Hg}}$	l	$ F_{\mathrm{Ri}} $	$ F_{\rm RiHg} $	ΔF	$4F_{ m Hg}$	1	$ F_{ m Ri} $	$ F_{\rm RiHg} $.4 F	$4F_{Hg}$
		001			1		$\overline{2}0i$!				30	!	
1	221	491	270	214	20	0	74	74	86	0	0	104	104	-275
2	194	147	-47	0	19	127	43	- 84	42	1	104	85	- 19	-268
3	455 944	516 311	61 67	-129	18	74	134	60 196	1	2	318	194	-124	- 96
5	147	294	147	- 94	17	199	325	-180	31 115	3	281	345	64 85	78
6	85	199	114	ŏ	$15 \\ 15$	120	321	201	147	5	120	203	- 83	112
7	224	134	-90	-119	· 14	60	0	-60	53	6	208	281	73	-19
8	180	180	0	-225	13	0	127	127	-117	7	85	180	95	49
9 10	216	159	- 57	181	12	336	303	- 33	-221	8	85	312	227	170
11	448	285	163	166	. 10	357	228	- 129	- 174	9	0 965	120	120	201
12	85	85	0	191	9	233	549	316	- 38 40	11	$\frac{205}{275}$	329	- 205 54	85
13	95	112	17	93	8	398	224	174	-5	12	441	393	-48	- 171
14	134	104	-30	0	7	258	443	185	-81	13	489	499	10	-125
10	127	43	- 84	1	6	95 490	0	- 95	- 44	14	294	262	-32	- 30
17	224	134	-90	75	4	420	208 134	-102 22	124	15	14	120	- 14	9 90
18	189	120	-79	0	3	332	969	637	265	17	95	85	-10	- 30
19	147	. 95	-52	-106	2	85	60	-25	73	18	224	95	-129	- 40
20	0	180	-180	- 147	1	358	248	-110	-125				_	
		101			i 1			_		91	0	40	ļ 	
20	112	95	-17	- 69		100	20/	1		21	60	208	60 148	1
19	74	0	-74	-91	: 0	400	320	-80	171	19	141	244	103	98
18	175	95 104	-80	-20	2	104	202	104	- 11	18	120	0	-120	71
17	180	164	- 16	111	3	60	ŏ	-60	-50^{2}	17	159	281	122	-48
10	60	95 104	95 	192	4	147	248	101	-172	16	164	0	- 164	171
14	281	463	182	19	5	323	405	82	-204	14	395	400 705		180
13	208	104	-104	-63	6	258	0	-258	-67	13	164	0	-164	- 05 35
12	493	794	301	- 36	8	43	104	-127	138	12	337	380	43	56
11	159	337	178	32	9	185	74	-111	173	11	180	244	64	-9
9	60	368	308	15 	10	417	347	-70	35	10	153	0	-153	- 42
8	112	104	-8	-249	11	127	0	-127	- 34	8	308 281	265	193	44
7	0	397	397	-226	12	325	351	26	5	7	164	104	-60	254
6	285	258	-27	- 37	13	221 197	212	- 9 79	60 20	6	120	285	165	136
5	0	112	112	164	15	185	95	-90	78	5	308	477	169	-73
43	297	120 262	- 177	214	16	60	74	14	-164	4	170	190	20	- 196
2	441	$\frac{262}{265}$	-176	6	17	43	85	42	-143	3	244	323	79	- 154
1	329	236	- 93	32		95	0	-95	-37	1	$\frac{224}{268}$	$\frac{228}{208}$	- 60 - 60	- 38
		107			19	U	0	0	- 98					0
Ω	202	515	199	145	İ		จิ๊ง	,			20.0	401		
ĩ	339	308	-31	145	01	104	30/			0	236	112	- 124	-87
2	194	85	-109	35	$\begin{vmatrix} 21\\ 20 \end{vmatrix}$	134	90 43	- 39 - 43	52 143	2	120	194	-213 -120	- 109
3	294	465	171	-186	19	170	43	-127	150	3	203	281	78	73
4	0	303	303	-295	18	194	216	22	58	4	104	104	0	235
0 6	190	208	134	-207	17	134	185	51	-51	ð	85	112	27	236
7	288	465	177	$-20 \\ 76$	16	459	597	138	-79	67	104	43	-61	97
8	95	104	9	39	10	109	331	- 159	- 25	8	0	40 95	- 42 95	- 57
9	255	268	13	-35	13	445	566	121	-43	9	Ő	43	43	9
10	248	424	176	-15	12	294	43	-251	-162	10	268	134	-134	37
12	493	794	301	106	11	443	585	142	-207	11	203	0	-203	- 35
13	199	0	-127 -199	171		251	0	- 251	- 95	12	180	0 199	- 180	144
14	291	447	156	26	ัช	112	112	119	103	14	236	74	-142	- 173
15	0	134	134	-106	7	74	141	67	178	15	365	199	-166	43
16 17	0	95	95	-128	6	175	0	-175	46	16	0	43	43	107
18	123	194 95	41	- 60	5	258	74	-184	-15	17	0	0	0	77
19	185	104	-81	-16	4 2	719 380	1200 494	481	51			50/		
					2	221	236	44 15	132	20	175	0	- 175	- 83
		201			1	407	337	-70	-107	19	127	74	-53	- 66
21	60	120	60	58						18	164	85	- 79	-11

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Table 1 (cont.)														
l	$ F_{ m Ri} $	$ F_{\mathrm{RiHg}} $ $\overline{5}0l$	ΔF	$4F_{ m Hg}$	1	$ F_{ m Ri} $	$ F_{\rm RiHg} $ 60l	∆F	$4F_{ m Hg}$	l	$ F_{ m Ri} $	$ F_{\rm RiHg} $	ΔF	$4F_{ m Hg}$
$ \begin{array}{r} 17 \\ 16 \\ 15 \\ 14 \\ 13 \\ 12 \\ 11 \\ 10 \\ 9 \\ 8 \\ 7 \\ 6 \\ \end{array} $	$147 \\ 385 \\ 0 \\ 153 \\ 391 \\ 244 \\ 141 \\ 337 \\ 212 \\ 315 \\ 147 \\ 203 \\$	0 358 0 300 127 0 473 175 120 281 159 307	$ \begin{array}{r} -147 \\ -27 \\ 0 \\ 147 \\ -264 \\ -244 \\ 332 \\ 162 \\ -92 \\ -34 \\ 12 \\ 104 \end{array} $		8 9 10 11 12 13 14 15 20 19	$74 \\ 95 \\ 0 \\ 233 \\ 0 \\ 306 \\ 0 \\ 95 \\ 371 \\ 0$	$\begin{array}{c} 60\\ 60\\ 251\\ 95\\ 221\\ 112\\ 281\\ 43\\ 104\\ \hline 700\\ 486\\ 0 \end{array}$	$ \begin{array}{r} -14 \\ 156 \\ 95 \\ -12 \\ 112 \\ -25 \\ 43 \\ 9 \\ 115 \\ 0 \\ \end{array} $	$ \begin{array}{r} -94 \\ -165 \\ -125 \\ 4 \\ 114 \\ 126 \\ 56 \\ -1 \\ -20 \\ -77 \\ \end{array} $	0 1 2 3 4 5 6 7 8 9 10 11	85 120 445 268 275 175 134 120 199 127 199 95	$185 \\ 141 \\ 399 \\ 141 \\ 251 \\ 43 \\ 185 \\ 60 \\ 159 \\ 228 \\ 120 \\ 95$	$100 \\ 21 \\ -46 \\ -127 \\ -24 \\ -132 \\ 51 \\ -60 \\ -40 \\ 101 \\ -79 \\ 0$	$ \begin{array}{r} -86 \\ -107 \\ -41 \\ 4 \\ -40 \\ -121 \\ -133 \\ -39 \\ 89 \\ 145 \\ 99 \\ 17 \end{array} $
5 4 3 2 1	0 288 194 216 153	104 347 297 104 208	$104 \\ 59 \\ 103 \\ -112 \\ 55$	-30 -200 -269 -159 36	18 17 16 15 14	306 153 175 43 127	$288 \\ 297 \\ 268 \\ 74 \\ 255 \\ 252 \\$	18 144 93 31 128	-93 -16 111 180 129	18 17 16	74 74 112	907 95 74 74	$21 \\ 0 \\ -38$	126 45 -42
0 1 2 3 4 5 6 7 8 9 10 11	$120 \\ 0 \\ 304 \\ 212 \\ 120 \\ 449 \\ 0 \\ 185 \\ 324 \\ 175 \\ 141 \\ 43$	$50l \\ 0 \\ 141 \\ 159 \\ 141 \\ 43 \\ 626 \\ 0 \\ 288 \\ 352 \\ 0 \\ 141 \\ 0$	$-120 \\ 141 \\ -145 \\ -71 \\ -77 \\ 177 \\ 0 \\ 103 \\ 28 \\ -175 \\ 0 \\ -43$	$146 \\ 108 \\ 17 \\ 8 \\ 101 \\ 180 \\ 129 \\ -34 \\ -174 \\ -182 \\ -80 \\ 12$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	285 194 0 134 104 398 281 561 199 43 153 221 0	$203 \\ 244 \\ 0 \\ 0 \\ 185 \\ 291 \\ 312 \\ 809 \\ 85 \\ 0 \\ 180 \\ 281 \\ 74$	$ \begin{array}{r} -82\\ 50\\ 0\\ -134\\ 81\\ -107\\ 31\\ 248\\ -114\\ -43\\ 27\\ 60\\ 74\end{array} $	$14 \\ -51 \\ -19 \\ 38 \\ 12 \\ -109 \\ -213 \\ -182 \\ -24 \\ 129 \\ 156 \\ 72 \\ 2$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$104 \\ 60 \\ 60 \\ 134 \\ 0 \\ 120 \\ 278 \\ 0 \\ 74 \\ 320 \\ 85 \\ 224 \\ 95 \\ 300$	$\begin{array}{c} 60\\ 60\\ 0\\ 43\\ 85\\ 74\\ 297\\ 0\\ 0\\ 306\\ 0\\ 255\\ 95\\ 236\\ \end{array}$	$ \begin{array}{r} -44 \\ 0 \\ -60 \\ -91 \\ 85 \\ -46 \\ 19 \\ 0 \\ -74 \\ -14 \\ -85 \\ 31 \\ 0 \\ -64 \end{array} $	$ \begin{array}{r} -57 \\ -11 \\ 15 \\ -40 \\ -133 \\ -160 \\ -67 \\ 79 \\ 158 \\ 117 \\ 26 \\ -6 \\ 46 \\ 96 \end{array} $
12 13 14 15 16	$134 \\ 194 \\ 170 \\ 85 \\ 43$	212 153 85 104 74	$78 \\ -41 \\ -85 \\ 19 \\ 31$	15 - 36 - 48 17 107	0 1 2 3	$185 \\ 153 \\ 0 \\ 0$	$70l \\ 112 \\ 85 \\ 127 \\ 60$	$-73 \\ -68 \\ 127 \\ 60$	33 118 133 21		104 120 0	141 90/ 95 147	37 - 25 - 147	53 - 75 - 178
20 19 18 17 16 15 14 13	120 224 0 185 147 275 437 216	$\overline{6}0l$ 228 74 170 0 329 495 0	-120 4 -15 -147 54 8 -216	-104 - 157 - 107 3 73 50 - 8 2	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	433 475 141 422 0 112 164 141 281 203	$\begin{array}{c} 473\\ 342\\ 439\\ 0\\ 153\\ 153\\ 0\\ 134\\ 85\\ 236 \end{array}$	$\begin{array}{r} 40 \\ -133 \\ 298 \\ -422 \\ 153 \\ 41 \\ -164 \\ -7 \\ -196 \\ 33 \end{array}$	$-135 \\ -202 \\ -134 \\ -13 \\ 46 \\ 15 \\ -30 \\ -9 \\ 74 \\ 133$	2 3 4 5 6 7 8 9	$ \begin{array}{r} 104 \\ 255 \\ 0 \\ 74 \\ 0 \\ 112 \\ 104 \\ 34 \\ \end{array} $	$ \begin{array}{c} 0\\ 208\\ 104\\ 74\\ 0\\ 0\\ 147\\ 159\\ \overline{10}.0 \end{array} $	$ \begin{array}{r} -104 \\ 47 \\ 104 \\ 0 \\ -112 \\ 43 \\ 25 \\ 4 \end{array} $	$-164 \\ -54 \\ 47 \\ 59 \\ 10 \\ -14 \\ 34 \\ 105$
13 12 11	216 185 208	332 0	-210 147 -208	100 190	13	203	230	20	133	16 15	$\begin{array}{c} 272\\ 332 \end{array}$	324 199	52 133	-42 - 133
10 9 8 7 6 5 4 3 2 1	$216 \\ 451 \\ 262 \\ 255 \\ 141 \\ 303 \\ 0 \\ 127 \\ 397 \\ 0$	95 599 297 294 74 258 104 180 488 60 60/	$121 \\ 148 \\ 35 \\ 39 \\ -67 \\ -45 \\ 104 \\ 53 \\ 97 \\ 60$	$ \begin{array}{r} 155 \\ -6 \\ -165 \\ -192 \\ -94 \\ 2 \\ -11 \\ -94 \\ -113 \\ 5 \\ \end{array} $	19 18 17 16 15 14 13 12 11 10 9 8	$\begin{array}{c} 43\\ 164\\ 482\\ 141\\ 127\\ 60\\ 0\\ 194\\ 0\\ 447\\ 413\\ 216\end{array}$	$\begin{array}{c} 43\\ 208\\ 352\\ 85\\ 0\\ 0\\ 104\\ 147\\ 203\\ 376\\ 589\\ 0\end{array}$	$\begin{array}{r} 0\\ 44\\ -130\\ -56\\ -127\\ -60\\ 104\\ -47\\ 203\\ -71\\ 186\\ -216\end{array}$	$74 \\ 31 \\ -3 \\ 34 \\ 110 \\ 131 \\ 42 \\ -102 \\ -180 \\ -133 \\ -25 \\ 26$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$85 \\ 0 \\ 112 \\ 127 \\ 194 \\ 147 \\ 112 \\ 258 \\ 120 \\ 255 \\ 244 \\ 134$	0 0 134 104 180 85 255 342 175 351 228 74	$ \begin{array}{r} -85\\ 0\\ 22\\ -23\\ -14\\ -62\\ 143\\ 84\\ 55\\ 96\\ -16\\ -60\end{array} $	$ \begin{array}{r} -137 \\ -56 \\ 25 \\ 32 \\ -16 \\ -33 \\ 35 \\ 138 \\ 169 \\ 84 \\ -49 \\ -119 \end{array} $
0 1 2	435 405 588	268 461 591	$-167 \\ 56 \\ 3$	$177 \\ 249 \\ 157$	6 5	210 300 112 185	509 0 60	-210 209 -112 -125	-16 -71 -33	$\begin{vmatrix} 2\\ 1 \end{vmatrix}$	194 74	$\begin{array}{c} 251 \\ 43 \end{array}$	57 31	-86 - 18
2 3 4 5 6 7	549 240 236 532 194	465 127 329 300 306	-84 -113 93 -232 112	$ \begin{array}{r} -4 \\ -91 \\ -57 \\ 9 \\ -2 \end{array} $	3 2 1	318 566 391 255	463 502 329 190	125 145 -64 -62 -65		0 1 2 3	275 127 0 0	10,00 255 127 104 60	$-20 \\ 0 \\ 104 \\ 60$	-5 - 61 - 106 - 65

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Table 1 (cont)

						-		(00/00-)						
l	$ F_{\rm Ri} $	$ F_{\rm RiHg} $	ΔF	$4F_{ m Hg}$	l	$ F_{ m Ri} $	$ F_{\rm RiHg} $	ΔF	$4F_{\mathrm{Hg}}$	l	$ F_{ m Ri} $	$ F_{\rm RiHg} $	ΔF	$4F_{ m Hg}$
		10,0	l				11,0	ı				11,0	!	
4	216	104	-112	46	9	159	43	-116	57	0	85	60	-25	23
5	199	164	- 35	135	8	258	422	164	10	1	95	0	-95	78
6	0	60	60	129	7	159	134	-25	6	2	0	95	95	52
					6	0	0	0	42	3	306	228	-78	4
					5	203	190	-13	51					
		11,0	l		4	190	112	-78	-22			$\overline{12},0i$		
13	0	43	43	-79	3	112	60	-52	-124	8	0	43	43	-7
12	112	180	68	26	2	60	0	-60	-157	7	Ō	127	127	_ 97
11	60	95	35	122	1	127	134	7	-87	6	95	12.	_ 05	107
10	240	95	- 145	129							00	0	- 55	- 107
10	210	50	- 14.7	120						Ð	0	0	0	- 48
					i					4	306	228	-78	1



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Fig. 4. Plot on Fig. l(b) of all peaks occurring on fullysharpened Patterson sections of soaked Dix crystal having peak height greater than half the average peak height. The crosses indicate the positions of the *AB* peaks of the mercury atoms in the soaked-Dix crystal as determined from the fully-sharpened Patterson sections.



Fig. 5. Comparison of the intensities of the 0k0 reflections between the soaked-Dix and normal crystals.

to photometer the reflections from Dix crystals and ribonuclease if a reliable conclusion is to be drawn about Hg positions that are now making $\frac{1}{4}$ of their full contribution to the intensities of the reflections. The following work briefly describes the procedure adopted.

The investigation was confined to the h0l reflections of the two crystals, whose diffraction patterns were recorded with a Buerger Precession camera which facilitated the measurement of intensities by means of a JCL—Walker Automatic Recording Microdensitometer.

Crystals of about the same size, of both types, were carefully selected for the intensity measurements, and the intensities were corrected for Lorentzpolarisation effects by Waser's function given in Volume II of the International Tables for Crystallography. The $F^2(h0l)$ values from the two crystals were independently scaled by the Wilson method (1942), with due allowance made for the presence of the Hg atoms in Dix's crystal. The Patterson difference map using $(\Delta |F|)^2$ as coefficients was calculated for spacings to 2.5 Å; it showed similarities to Fig. 1(a) and was also uninterpretable. A smoothing function was then used which terminated the series at 6 Å, and Fig. 1(c) shows the resulting Patterson difference map projected on (010).

The similarity between Figs. 1(b) and (c) is at once apparent, though the peaks on 1(c) show different relative heights compared with those on 1(b). The pattern however is again interpretable in terms of 2 pairs of atoms related through a centre. The x and z coordinates of the A and B sites, which can only be approximate, are given in the table below together with the corresponding x and z psotitions determined from the analysis of the 3-dimensional Patterson sections of the soaked crystals.

	A (s	ite)	B (site)			
Crystals	\boldsymbol{x}	z	\boldsymbol{x}	z		
Unsoaked Dix crystal	0.12	0.08	0.78	0.18		
Soaked Dix crystal	0.153	0.075	0.806	0.150		

The agreement is sufficiently close to show that the mercury atoms occupy the same sites in both the unsoaked and soaked Dix crystal. The Δy values for the A and B sites in Dix's crystal have not been

determined, but it is unlikely that they will be any different from those already found for the soaked ones.



Fig. 6. Plot of the fractional change in intensity, $\varphi \Delta I = 2\{(\Delta |F|)^2 |I_{\rm Rl}\}^{\frac{1}{2}}$, for Dix crystal—normal crystal, versus $\sin \theta / \lambda$. The initial fall in $\varphi \Delta I$ is due to the mercury atoms in the Dix crystal and the increase for $\sin \theta / \lambda > 0.14$ is attributed to a shift in the molecules of one crystal relative to those of the other.

Fig. 6 shows a plot of $\varphi \Delta I = 2 \{ (\Delta |F|)^2 / I_{\rm Ri} \}^{\frac{1}{2}}$ against $\sin \theta / \lambda$ (Crick and Magdoff, 1956). The initial decrease in the function is due to the effect of the heavy atoms, but the small increase towards higher $\sin \theta / \lambda$ values indicates that there must be a slight shift of the molecules in the Dix crystals relative to those in the metal-free ones. This is in agreement with the previous findings of Bernal, Carlisle and Rosemeyer (1959) for the soaked crystals.

4. Conclusions

This work has shown how the positions of heavy atoms in protein crystals, particularly when there is more than one per asymmetric unit, may be found by using fully sharpened three-dimensional Patterson functions, and this is all the more important where there is a likelihood that isomorphism is not strictly obeyed owing to the relative movement of the molecules in the two crystals.

The structure factors of the mercury atoms in the soaked crystal have now been calculated to spacings of 2A, leading to the synthesis of two three-dimensional Fourier distributions for the metal-free crystal incorporating some 60% of the reflections in each case to (a) the 6Å region, and (b) the 2.5Å region. From this it has been possible to gain some knowledge of the configuration of the single polypeptide chain in this molecule and also the positions of the four cystine sulphurs in it.

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